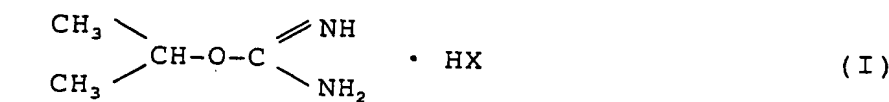


AMENDMENTS TO THE SPECIFICATION

1. On page 3, please replace 2<sup>nd</sup> paragraph with the following paragraph:

The above-mentioned problem have been solved by reacting cyanamide and isopropyl alcohol in the presence of sulfuric acid to thereby obtain O-isopropyl-isourea hydrogen sulfate at a high yield. The sulfate can be obtained by neutralizing the above hydrogen sulfate with an alkali metal hydroxide-;

O-isopropyl-isourea hydrogen sulfate or sulfate represented by the formula (I):



wherein X represents HSO<sub>4</sub> or 1/2 SO<sub>4</sub>.

2. On page 5, please replace 1<sup>st</sup> paragraph with the following paragraph:

Furthermore, when ~~equal moles~~ 1 mole of the resultant O-isopropyl-isourea sulfate and ½ mole of sulfuric acid are reacted to form O-isopropyl-isourea hydrogen sulfate. The resultant O-isopropyl-isourea can be isolated by concentrating to dryness or by dispersing in a poor solvent.

3. On page 6, please replace 2<sup>nd</sup> paragraph with the following paragraph:

To a 500 ml four-necked flask provided with a stirrer, a thermometer and a starting material charge device, 64.29g of distilled water and 64.29g of conc. sulfuric acid (concentration 98.1%, 0.64 mol) were charged, while cooling, and 200g of O-isopropyl-isourea sulfate (purity 97.2%, ~~0.64~~ 1.29 mol) was added thereto, while stirring. The reaction solution was concentrated under vacuo and the concentrated mixture was dispersed in hexane, followed by separating the precipitated crystal by filtration under vacuo. Thus, 244.49 of O-isopropyl-isourea hydrogen sulfate was obtained by drying at room temperature under vacuo.